Effect of Polyhalite application on the yield, quality and shelf life of green pepper in Hainan province, China

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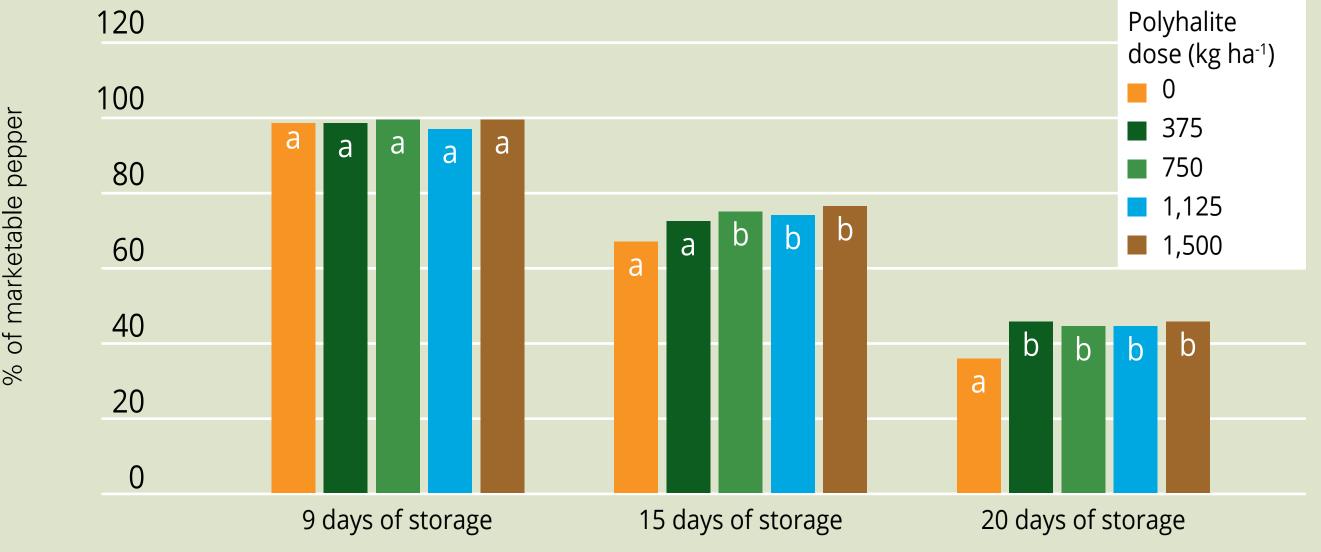
Introduction

Green pepper is one of the most important cash crops in Hainan province, representing the prominence that China has gained in the tropical crop industry (Zu et al., 2014). The soil pH in approximately 50% of pepper fields in Hainan province is below 5.5, falling below the suitable green pepper soil pH range of 5.5-7 (Yang et al., 2009). Growing green peppers on low pH soil often results in poor growth, nutrient deficiencies (potassium - K, calcium - Ca and magnesium - Mg), low yield and poor quality.

Polyhalite, K₂Ca₂Mg(SO₄)₄•2H₂O (George, 1931; Luca, 2005), marketed as Polysulphate by ICL Fertilizers, is a naturally occurring mineral which is certified as an organic fertilizer, containing 48% SO₃, 14% K₂O, 6% MgO and 17% CaO, all of which are fully soluble. This new four-in-one fertilizer is abundant in Ca and Mg which makes it the ideal fertilizer for the tropical agricultural regions of China.

Results and discussion

Fig. 1 Effect of Polyhalite application on percentage of marketable pepper after harvest



Objective

The objective of this study was to quantify the effects of Polyhalite on green pepper production in Hainan province including: yield, quality and shelf life.

Materials and methods

Properties of soil

The soil is a sandy loam with a pH value of 5. The properties of the soil were: organic matter 14.12 g kg⁻¹, alkali-hydrolyzed nitrogen 96.45 mg N kg⁻¹, available phosphorus 13.21 mg P₂O₅ kg⁻¹, available potassium 65.12 mg K₂O kg⁻¹, exchangeable calcium 763.25 mg Ca kg⁻¹, exchangeable magnesium 115.36 mg Mg kg⁻¹.

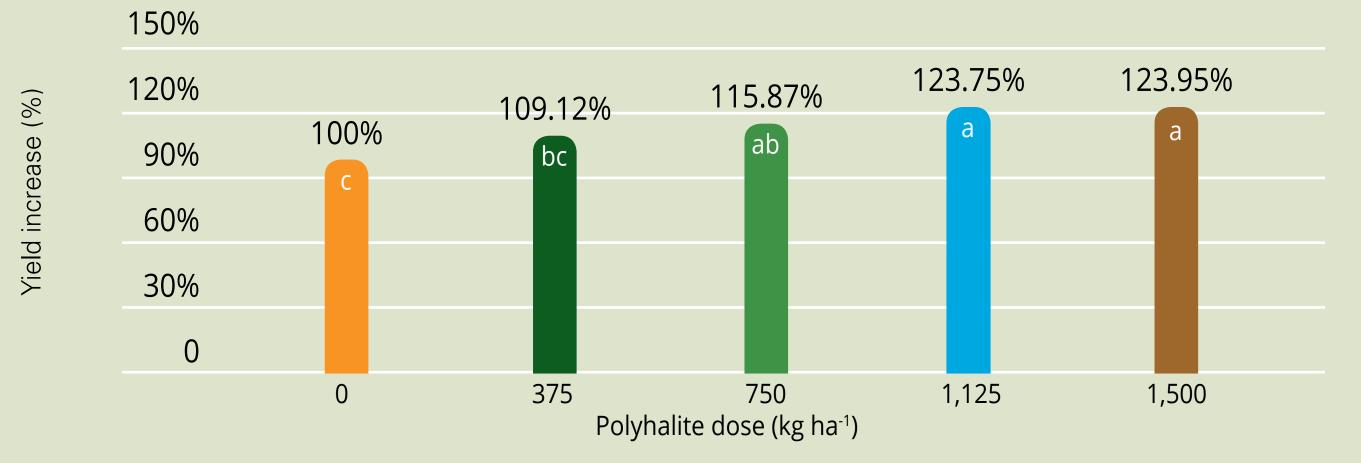
Experimental design

The field experiment was conducted at a farm in Wenchang city, Dongyang county, Hainan province. The experiment was laid out in a complete randomized block design with three replications.

Different letters within columns indicate statistically significant differences

There was no significant difference in the percentage of marketable pepper among all treatments after 9 days of storage. As the storage time was increased from nine days to 15 and then 20 days, a significantly higher percentage of marketable pepper was achieved in treatments that contained Polyhalite, due to improved shelf life.

Fig. 2 Effect of Polyhalite application on pepper yield



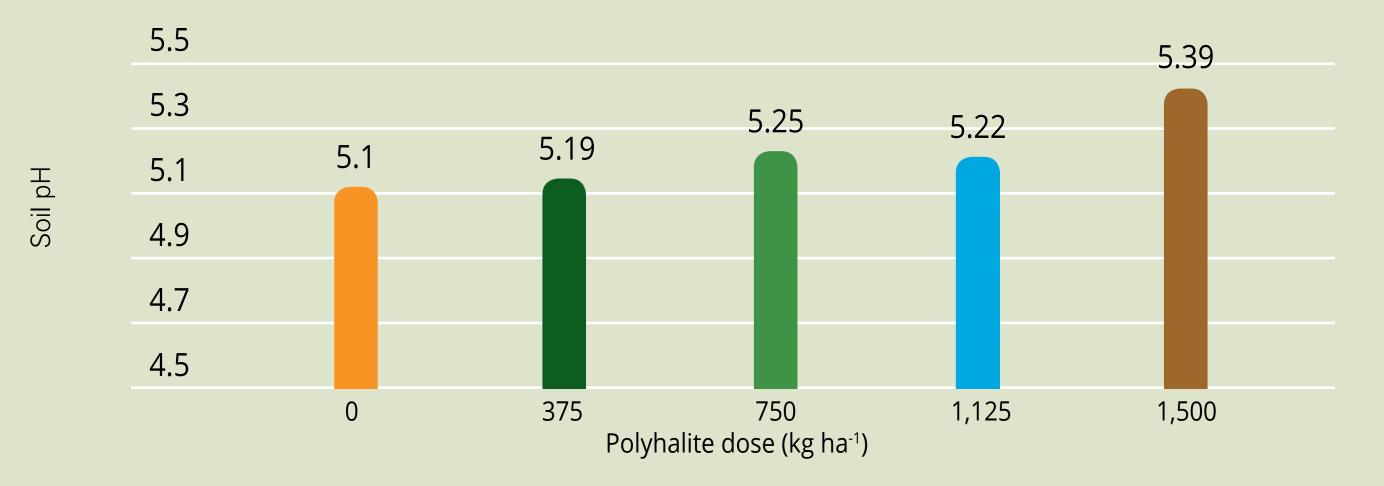
Different letters within columns indicate statistically significant differences

Comparing Polyhalite treatments with the control, the yields of green pepper increased significantly by up to 24% at the highest dose (1,500 kg Polyhalite ha⁻¹)

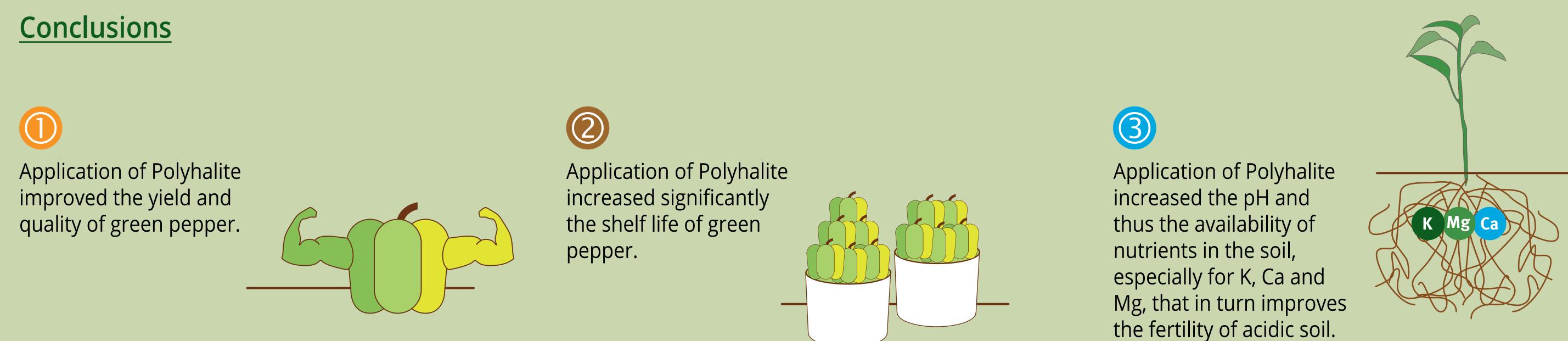
- Control: Farmers' traditional practice: 1,125 kg ha⁻¹ of compound fertilizer (15-15-15) applied as base-fertilizer; topdressing of 375 kg ha⁻¹ of compound fertilizer at fruit stage;
- Treatment 1: Farmer practice + 375 kg ha⁻¹ Polyhalite
- Treatment 2: Farmer practice + 750 kg ha⁻¹ Polyhalite
- Treatment 3: Farmer practice + 1,125 kg ha⁻¹ Polyhalite
- Treatment 4: Farmer practice + 1,500 kg ha⁻¹ Polyhalite

The area of each plot was 20 m² (1.43 m X 14 m); plant and row spacing 25 cm × 40 cm. Control of weeds, insects, and diseases were done according to local standard prevention recommendations.

Fig. 3 Effect of Polyhalite application on soil pH after pepper harvest



Comparing Polyhalite treatments with the control, the application of Polyhalite increased soil pH from 5.1 to 5.39.



References: Zu, C., Li, Z.G., Yang, J.Y., Yu, H., Sun, Y., Tang, H.L., Yost, R., Wu, H.S., (2014) Acid Soil Is Associated with Reduced Yield, Root Growth and Nutrient Uptake in Black Pepper (*Piper nigrum* L.). Agricultural Sciences, 5, 466-473. // Yang, J.F., Xing, G.Y., Sun, Y., Wang, H., Wu, H.S. and Zheng, W.Q. (2009) Analysis and Assessment of Soil Chemical Fertility in Typical Black Pepper Gardens in Hainan. Journal of Tropical Crops, 30, 1291-1294. // George, F.R., (1931) Recovery of potassium sulphate from Polyhalite. US Patent 1,812,497. // Luca Bindi. (2005) Reinvestigation of polyhalite, K₂Ca₂Mg(SO₄)₄-2H₂O. Acta Cryst, E61, i135–i136.



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